

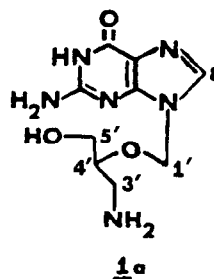
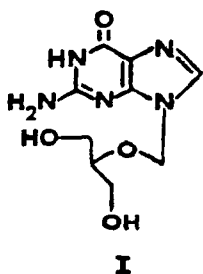
A NOVEL METHOD FOR THE SYNTHESIS OF
9-[[2-HYDROXY-1-(AMINOMETHYL)ETHOXY]METHYL]GUANINE
- A POTENTIAL ANTIVIRAL AGENT¹

Mao-Chin Liu², Sandra Kuzmich and Tai-Shun Lin^{*}

Department of Pharmacology and Comprehensive Cancer Center
Yale University School of Medicine, New Haven, Connecticut 06510, USA

9-[[2-Hydroxy-1-(aminomethyl)ethoxy]methyl]guanine (**1a**), an amino analogue of 9-[[2-hydroxy-1-(hydroxymethyl)ethoxy]methyl]guanine (**I**) which is a potent antiviral agent, has been synthesized via a multistep-synthesis.

Recently, a novel guanine acyclic nucleoside analogue, 9-[[2-hydroxy-1-(hydroxymethyl)ethoxy]methyl]guanine (**I**) (BIOLF-62, 2'NDG, DHPG) has been reported independently by several laboratories³⁻⁷ to have potent antiviral activity against both herpes simplex virus type 1 and 2 *in vitro* as well as *in vivo* at concentrations well below cytotoxic levels. Now we wish to report the synthesis of the amino analogue of **I**. Since the amino function is sterically and electronically similar to that of the hydroxy group, the modification of the acyclic sugar moiety with an amino group may lead to compounds with high antiviral activity and low host cytotoxicity. Furthermore, the amino analogue **1a** can be readily converted to the corresponding hydrochloride, acetate or other salts. These salts should markedly enhance the water solubility, a factor critical for the formulation, which often limits the usefulness of new potential antiviral agents. Recently, Martin et al.⁸, have also reported the synthesis of **1a** by a completely different method.

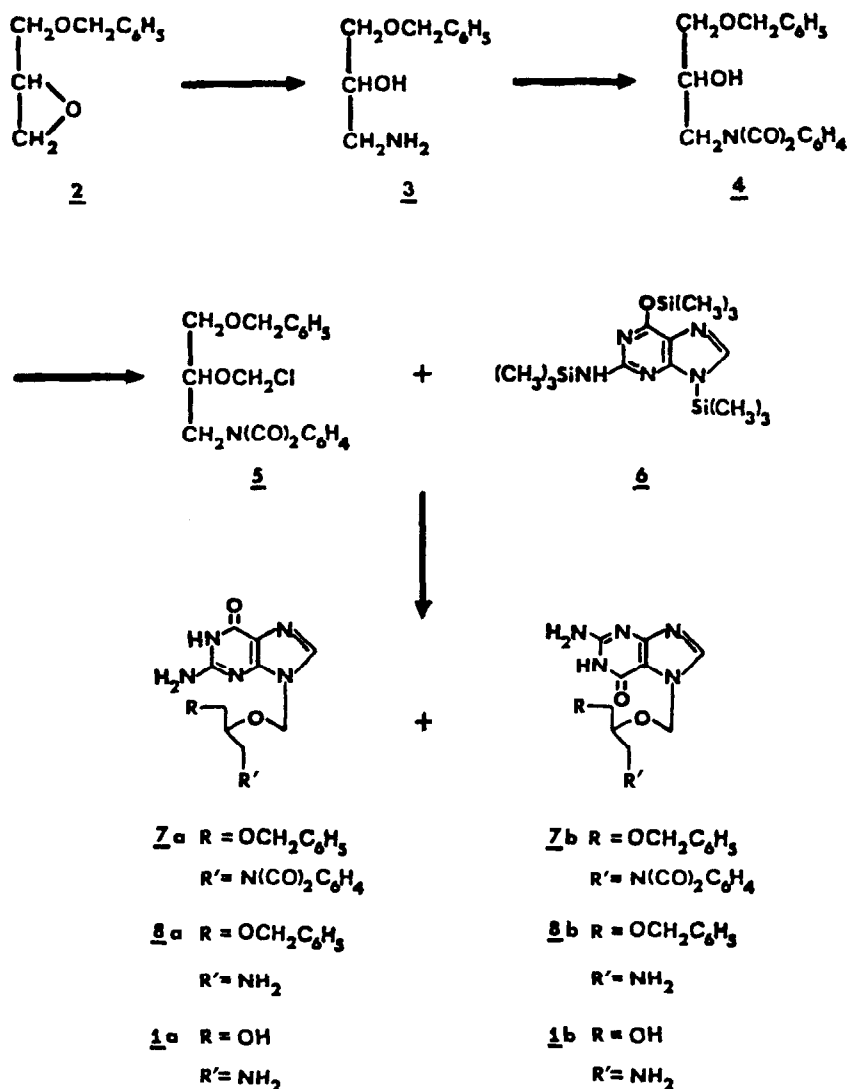


The ether **2** which was prepared by a reported literature procedure⁹, was stirred with concentrated (28%) NH_4OH at room temperature for 3 days to afford the desired 1-amino-3-benzyloxy-2-propanol (**3**) in 72% yield: mp 76°C ; NMR (CDCl_3) δ 2.38 (s, 3H, $\text{H}_2\text{N}-1$ and $\text{OH}-2$, D_2O exchangeable), 2.80 (d, 2H, CH_2N), 3.49 (d, 2H, OCH_2C), 3.80 (m, 1H, CH), 4.54 (s, 2H, ArCH_2O), 7.32 (m, 5H, C_6H_5). Compound **3** was reacted with phthalic anhydride to afford N-(3-benzyloxy-2-hydroxypropyl) phthalimide (**4**) (65%): mp $79-80^\circ\text{C}$; NMR (CDCl_3) δ 2.81 (s, 1H, OH, D_2O exchangeable), 3.51 (d, 2H, OCH_2C), 2.83 (d, 2H, CH_2N), 3.97 (m, 1H, CH), 4.52 (s, 2H, ArCH_2O), 7.30 (m, 5H, C_6H_5), 7.68 (m, 4H, C_6H_4). The key intermediate, the chloromethyl ether (**5**), was synthesized in almost quantitative yield by treatment of **4** with paraformaldehyde and dry HCl in 1,2-dichloroethane at $0-5^\circ$ for 2 hr. and after drying with CaCl_2 to remove the water formed, the reaction was continued for another 2 hr. Chloride (**5**) was used for the next step without further purification: NMR (CDCl_3) δ 3.62 (d, 2H, OCH_2C), 3.87 (d, 2H, CH_2N), 4.28 (m, 1H, CH), 4.56 (s, 2H, ArCH_2O), 5.46 (s, 2H, OCH_2Cl), 7.26 (m, 5H, C_6H_5), 7.72 (m, 4H, C_6H_4).

Alkylation of tris(trimethylsilyl)guanine (**6**) with (**5**) in the presence of triethylamine in toluene gave a mixture of the protected guanine acyclic nucleoside **7a** and its isomer **7b** in 80% yield. The ratio of the two isomers could be varied depending on the reaction conditions used. For example, the ratio (by comparing the integration of NMR spectra of the mixture) of **7a** over **7b** was 3:2 when the reaction mixture was refluxed for 48hr. However, when the reaction mixture was stirred first at room temperature for 24 hr. and then refluxed for another 24 hr. it yielded a mixture of **7a** and **7b** in a ratio of 4:1. The two isomers, **7a** and **7b**, were separated by crystallizing from DMSO (**7a** was more soluble), and purified by silica gel column chromatography (CHCl_3 -EtOH, 2:1): **7a**, mp $226-228^\circ\text{C}$; NMR ($\text{DMSO}-d_6$) δ 2.50-3.68 (m, 4H, H-3' and H-5'), 4.12 (m, 1H, H-4'), 4.50 (s, 2H, ArCH_2O), 5.30 (q, 2H, H-1'), 6.24 (s, 2H, NH_2 , D_2O exchangeable), 7.31 (m, 5H, C_6H_5), 7.57 (s, 1H, H-8), 7.65-7.74 (m, 4H, C_6H_4), 10.14 (s, 1H, NH-1, D_2O exchangeable); **7b**, mp $275-277^\circ$, NMR ($\text{DMSO}-d_6$) δ 3.49-3.68 (m, 4H, H-3' and H-5'), 4.21 (m, 1H, H-4'), 4.51 (s, 2H, ArCH_2O), 5.50 (q, 2H, NCH_2O), 5.93 (s, 2H, NH_2 , D_2O exchangeable), 7.31 (m, 5H, C_6H_5), 7.65-7.75 (m, 4H, C_6H_4), 7.87 (s, 1H, H-8), 10.40 (s, 1H, NH-1, D_2O exchangeable). The phthaloyl protective groups of **7a** and **7b** were removed with hydrazine in EtOH to yield **8a** (65%): mp $100-102^\circ\text{C}$; NMR ($\text{DMSO}-d_6$) δ 2.51 (d, 2H, H-3'), 3.45 (d, 2H, H-5'), 3.70 (m, 1H, H-4'), 4.10 (s, 2H, NH_2-3' , D_2O exchangeable), 4.40 (s, 2H, ArCH_2O), 5.38 (q, 2H, H-1'), 6.50 (s, 2H, NH_2-2), 7.18 (m, 5H, C_6H_5), 7.74 (s, 1H, H-8), and **8b** (60%): 180°C (dec); NMR ($\text{DMSO}-d_6$) δ 2.70 (d, 2H, H-3'), 3.41 (d, 2H, H-5'), 3.80 (m, 1H, H-4'), 4.30 (s, 2H, NH_2-3' , D_2O exchangeable), 4.40 (s, 2H, ArCH_2O), 5.62 (q, 2H, H-1'), 6.42 (s, 2H, NH_2-2 , D_2O exchangeable), 7.30 (m, 5H, C_6H_5), 8.10 (s, 1H, H-8).

Debenzylation of **7a** and **7b** were unusually difficult. However, the final products **1a** and **1b** were obtained by debenzylation of **7a** and **7b**, respectively, through hydrogenation in methanol using PdCl_2 as a catalyst. The debenzylation of **7b** was more difficult than that of **7a** with **7b** requiring 3 days to complete the reaction (followed by NMR). After removal of the catalyst and the solvent, the final products were obtained as hydrochloride salts. The products were recrystallized from EtOH- H_2O : **1a.HCl** (65%); mp $158-160^\circ\text{C}$; NMR δ 2.84-3.49 (m, 4H, H-3' and H-5'), 3.89 (m, 1H, H-4'), 5.05 (s, 1H, OH, D_2O exchangeable), 5.45 (q, 2H, H-1'), 6.74 (s, 2H, NH_2-2 , D_2O exchangeable), 7.86 (s, 1H, H-8), 8.05 (s, 3H, NH_3-3' , D_2O exchangeable), 10.89 (s, 1H, NH-1, D_2O exchangeable); **1b.HCl** (52%), mp $208-210^\circ\text{C}$ (dec); NMR

(DMSO- d_6) δ 2.84-3.49 (m, 4H, H-3' and H-5'), 3.88 (m, 1H, H-4'), 5.10 (s, 1H, OH, D_2O exchangeable), 8.12 (s, 1H, H-8), 8.15 (s, 3H, NH_3^+-3' , D_2O exchangeable). The free base **1a** and **1b** were obtained by adjusting the aqueous solution of **1a.HCl** and **1b.HCl** with 2N NaOH to pH 8.8, and then recrystallized from water: **1a**:mp 148-150°C; UV (H_2O), λ_{max} 271 nm (sh); λ_{min} 253 nm; λ_{min} 221 nm; NMR (DMSO- d_6) δ 2.45 (d, 2H, H-3'), 3.35 (d, 2H, H-5'), 3.53 (m, 1H, H-4'), 4.10 (s, 3H, OH and NH_2^+-3' , D_2O exchangeable), 5.35 (q, 2H, H-1'), 6.51 (s, 2H, NH_2^+-2 , D_2O exchangeable), 7.74 (s, 1H, H-8); **1b**:mp 160-162°C, UV (H_2O) λ_{max} 285 nm, λ_{min} 261 nm, λ_{242} nm (sh), NMR (DMSO- d_6) δ 2.60 (d, 2H, H-3'), 3.34 (d, 2H, H-5'), 3.60 (m, 1H, H-4'), 4.25 (s, 3H, OH and NH_2^+-3' , D_2O exchangeable), 5.60 (q, 2H, H-1'), 6.41 (s, 2H, NH_2^+-2 , D_2O exchangeable), 8.02 (s, 1H, H-8).



The structural assignments of the isomers **1a** and **1b** were made on the basis of the comparison of their spectroscopic and chemical properties with those of the similar known compounds.^{5,10,11} The UV spectra of **1a** and **1b** were similar to those observed for 9-methyl and 7-methylguanine¹⁰ respectively. The NMR chemical shift of H-8 of **1b** (8.02 δ) was more downfield than that of **1a** (7.74 δ) and the same phenomenon was also observed for the similar compounds of N²-acetyl-9-[[1,3-bis(benzyloxy)-2-propoxy]methyl]guanine and N²-acetyl-7-[[1,3-bis(benzyloxy)-2-propoxy]methyl]guanine⁵. The relative stability of **1a** and **1b** to acid catalyzed hydrolysis was compatible to those of guanosine and its 7-isomer¹¹.

All new compounds gave satisfactory elemental analysis for carbon, hydrogen and nitrogen. The biological and biochemical studies of **1a** and its (R) and (S) optical isomers are underway, and the results will be reported elsewhere.

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